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8α -Hydroxy- 11α , 13-dihydrozaluzanin C

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Abstract

In the structure of the title compound [3a,4,5,6,6a,7,8,9,9a,9b-decahydro-4,8-dihydroxy-3-methyl-6,9-bis(methylene)azuleno [4,5-b] furan-2(3H)-one, $C_{15}H_{20}O_4$], the secondary hydroxyl groups participate in intermolecular hydrogen bonds both as donors and acceptors.

Comment

The title molecule, (I), was isolated previously from Ainsliaea fragrans Champ. as an oily substance (Bohlmann & Chen, 1982). Recently, we obtained the same compound from Centaurea scoparia Sieb. as colourless crystals (Helal et al., 1997). The X-ray crystal and molecular structure determinations confirm the relative stereochemistry unambiguously.

The title compound has a cyclopentane ring cis-fused and a γ -lactone ring trans-fused to a cycloheptane ring (Fig. 1). The γ -lactone ring has a half-chair conformation with a pseudo-twofold axis running through the C12 carbonyl atom. The cyclopentane ring exhibits a distorted half-chair conformation with the C4 atom on the approximate twofold axis. The torsion angles of the cycloheptane ring have C_2 symmetrical values and the ring is therefore in a twist-chair conformation. The pseudo-twofold axis bisects the C7—C8 bond and passes through C1.

The secondary hydroxyl groups participate in intermolecular hydrogen bonds both as donors and acceptors. Atom O3 donates a hydrogen bond to O8 and also ac-

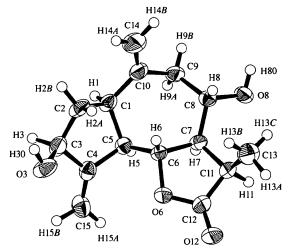


Fig. 1. The molecular structure of (I) showing 50% probability displacement ellipsoids.

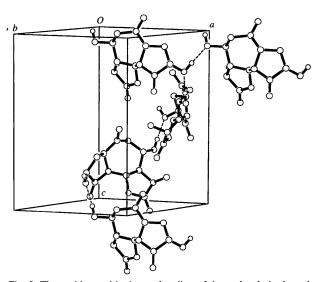


Fig. 2. The packing and hydrogen bonding of the molecule in the unit cell. All H atoms, except for those of the hydroxyl groups, have been omitted for clarity.

cepts a hydrogen bond from O8 (Table 2). Thus, the molecules are linked together by the latter hydrogen bond to form chains running around the 3_2 axes with each chain connected by the former hydrogen bond to form a network (Fig. 2).

Experimental

The title compound was isolated as reported previously (Helal *et al.*, 1997). Crystals suitable for X-ray analysis were recrystallized from chloroform (m.p. 442 K).

Crystal data

 $C_{15}H_{20}O_4$ $M_r = 264.32$ Mo $K\alpha$ radiation $\lambda = 0.71069 \text{ Å}$

Trigonal	Cell parameters from 25
P3 ₂	reflections
a = 9.221 (4) Å	$\theta = 10.2 - 12.0^{\circ}$
c = 14.006 (6) Å $V = 1031.4 (9) \text{ Å}^3$	$\mu = 0.092 \text{ mm}^{-1}$
$V = 1031.4(9) \text{ Å}^3$	T = 293 K
Z = 3	Prismatic
$D_x = 1.277 \text{ Mg m}^{-3}$	$0.34 \times 0.31 \times 0.17 \text{ mm}$
D_m not measured	Colourless

Data collection

Rigaku AFC-7R diffractom-	$R_{\rm int}=0.029$
eter	$\theta_{\rm max} = 27.48^{\circ}$
ω –2 θ scans	$h = -10 \rightarrow 10$
Absorption correction: none	$k = 0 \rightarrow 11$
1780 measured reflections	$l = 0 \rightarrow 18$
1636 independent reflections	3 standard reflections
1150 reflections with	every 150 reflections
$I > 2\sigma(I)$	intensity decay: 0.68%

Refinement

Refinement on F^2	$(\Delta/\sigma)_{\text{max}} = -0.011$
$R[F^2 > 2\sigma(F^2)] = 0.045$	$(\Delta/\sigma)_{\text{max}} = -0.011$ $\Delta\rho_{\text{max}} = 0.156 \text{ e Å}^{-3}$
$wR(F^2) = 0.121$	$\Delta \rho_{\min} = -0.161 \text{ e Å}^{-3}$
S = 0.926	Extinction correction: none
1636 reflections	Scattering factors from
172 parameters	International Tables for
H atoms not refined	Crystallography (Vol. C)
$w = 1/[\sigma^{2}(F_{o}^{2}) + (0.1P)^{2}]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$	
where $P = (F_o^2 + 2F_c^2)/3$	

Table 1. Selected geometric parameters (Å, °)

	3	F (12)	,
C1—C2 C1—C5 C1—C10	1.520 (6) 1.566 (5) 1.510 (7)	C2—C3 C9—C10	1.513 (7) 1.519 (6)
C2—C1—C5 C2—C1—C10 C5—C1—C10 O3—C3—C2 O3—C3—C4 C2—C3—C4 C1—C5—C4 C1—C5—C6 C4—C5—C6 O6—C6—C5 O6—C6—C7	103.3 (3) 115.9 (4) 116.5 (3) 114.5 (4) 110.0 (3) 105.1 (3) 102.3 (3) 108.6 (3) 116.1 (3) 110.8 (3) 103.4 (3)	C5—C6—C7 C6—C7—C8 C6—C7—C11 C8—C7—C11 O8—C8—C7 O8—C8—C9 C7—C8—C9 C7—C11—C12 C7—C11—C13 C12—C11—C13	114.4 (3) 114.5 (3) 103.1 (3) 117.3 (3) 105.4 (3) 109.9 (3) 112.5 (3) 102.2 (3) 117.1 (3) 109.4 (4)
06—C6—C7—C11 06—C12—C11—C7 C1—C2—C3—C4 C1—C5—C4—C3 C1—C5—C6—C7 C1—C10—C9—C8 C2—C1—C5—C4 C2—C3—C4—C5 C3—C2—C1—C5	-32.3 (4) -12.4 (5) 28.8 (4) -17.3 (4) -91.0 (4) -89.4 (5) 34.6 (4) -6.6 (4) -39.7 (4)	C5—C1—C10—C9 C5—C6—C7—C8 C6—O6—C12—C11 C6—C5—C1—C10 C6—C7—C8—C9 C6—C7—C11—C12 C7—C6—O6—C12 C7—C8—C9—C10	38.0 (5) 78.6 (4) -8.6 (5) 39.6 (4) -54.7 (4) 26.9 (4) 25.9 (4) 71.6 (4)

Table 2. Hydrogen-bonding geometry (Å, °)

D — $H \cdot \cdot \cdot A$	<i>D</i> H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	<i>D</i> H· · · <i>A</i>
O3—H30· · ·O8 ⁱ	0.917	1.873	2.739 (5)	156.5
O8—H80· · ·O3 ⁱⁱ	0.880	1.912	2.790(5)	175.3
Symmetry codes: (i)	1 + x, y, z; (i)	i) $1 - v, x -$	$-v.z - \frac{1}{2}$.	

All H atoms were obtained from difference Fourier synthesis but not refined. Isotropic displacement parameters of the H atoms were set equal to $1.05U_{\rm eq}$ of the bonded non-H atom.

Data collection: MSC/AFC Diffractometer Control Software (Molecular Structure Corporation, 1995a). Cell refine-

ment: MSC/AFC Diffractometer Control Software. Data reduction: TEXSAN (Molecular Structure Corporation, 1995b). Program(s) used to solve structure: SIR92 (Altomare et al., 1994). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: ORTEPII (Johnson, 1976) and CHARON (Lauher, 1989). Software used to prepare material for publication: TEXSAN.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: NA1327). Services for accessing these data are described at the back of the journal.

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Diethyl 3,7-Hypoxanthyldiacetate

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Abstract

The title compound, diethyl 6-oxo-6,7-dihydro-3H-purine-3,7-diacetate, $C_{13}H_{16}N_4O_5$, was obtained by direct alkylation of hypoxanthine using ethyl bromoacetate, in the presence of potassium carbonate; the two alkyl substituents are attached to the heterocycle at positions N3 and N7. Both side chains avoid steric hindrance with the heterocycle, as demonstrated by the torsion angles C2—N3—C15—C16 -68.9 (2) and C8—N7—C10—C11 106.8 (2)°.